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## Phosphorus, Sulfur, and Silicon and the Related Elements

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### Novel Compounds Formed from Reactions of $[(\epsilon^5\text{-Me}_5\text{C}_5)\text{MCl}_2]_2$ (M = Ru, Rh) with Vinylidiphenylphosphine and Allyldiphenylphosphine

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NOVEL COMPOUNDS FORMED FROM REACTIONS OF  $[(\eta^5\text{-Me}_5\text{C}_5)\text{MCl}_2]$  ( $\text{M} = \text{Ru, Rh}$ ) WITH VINYLDIPHENYLPHOSPHINE  
AND ALLYLDIPHENYLPHOSPHINE

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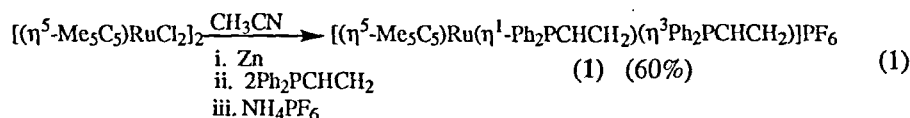
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**Abstract** The complexes  $[(\eta^5\text{-Me}_5\text{C}_5)\text{Ru}(\text{R}_3\text{P})_2(\text{NCCH}_3)]^+ \text{PF}_6^-$ , where  $\text{R}_3\text{P}$  is vinylidiphenylphosphine and allyldiphenylphosphine, readily dissociate  $\text{CH}_3\text{CN}$  to form  $\eta^3$ -phosphaallyl and  $\eta^3$ -homophosphaallyl complexes (**1** and **2**) respectively. While **1** is conformationally rigid in solution, **2** undergoes inversion by a dissociative process in solution. The analogous complex  $[(\eta^5\text{-Me}_5\text{C}_5)\text{Rh}(\text{Ph}_2\text{PCHCH}_2)_2\text{Cl}]^+ \text{PF}_6^-$  undergoes a novel Michael type addition of the methyl CH bonds across the vinyl functionalities of the coordinated phosphine to produce chelating 1,2- and 1,3-*bis*-(diphenylphosphinopropyl) trimethylcyclopentadienides (**3** and **4**) respectively. In contrast, with  $\text{Ph}_2\text{PCH}_2\text{CHCH}_2$  the classical phosphine complex  $[(\eta^5\text{-Me}_5\text{C}_5)\text{Rh}(\text{Ph}_2\text{PCH}_2\text{CHCH}_2)_2\text{Cl}]^+ \text{PF}_6^-$  (**5**) is formed. These new complexes have been characterized by a variety of NMR techniques and by single crystal X-ray crystallography.

**Key Words** phosphaallyl, NMR spectroscopy, X-ray crystallography, CH Michael additions.

We previously reported<sup>1</sup> that  $[(\eta^5\text{-C}_5\text{H}_5)\text{Ru}(\text{Ph}_2\text{PCHCH}_2)_2(\text{CH}_3\text{CN})]^+ \text{PF}_6^-$  reversibly dissociates  $\text{CH}_3\text{CN}$  to form the  $\eta^3$ -phosphaallyl complex  $[(\eta^5\text{-C}_5\text{H}_5)\text{Ru}(\eta^1\text{-Ph}_2\text{PCHCH}_2)(\eta^3\text{-Ph}_2\text{PCHCH}_2)]^+ \text{PF}_6^-$  and showed that the  $\eta^1$ -to- $\eta^3$  conversion is endothermic and entropy driven. Since  $\eta^5\text{-Me}_5\text{C}_5^-$  is a better donor than  $\text{C}_5\text{H}_5^-$ , the metal in complexes of the former should be more electron rich than in complexes of the latter. An electron rich metal center would be expected to facilitate the  $\eta^1$ -to- $\eta^3$  conversion. Accordingly, the  $\eta^5\text{-Me}_5\text{C}_5$

phosphaallyl complex was synthesized in high yield *via* reaction 1. That the  $\eta^3$ -phosphaallyl complex was formed in this



reaction rather than  $[(\eta^5\text{-Me}_5\text{C}_5)\text{Ru}(\text{Ph}_2\text{PCHCH}_2)_2(\text{CH}_3\text{CN})]^+ \text{PF}_6^-$  was established by  $^{31}\text{P}\{^1\text{H}\}$  NMR spectroscopy and X-ray crystallography (Figure 1).

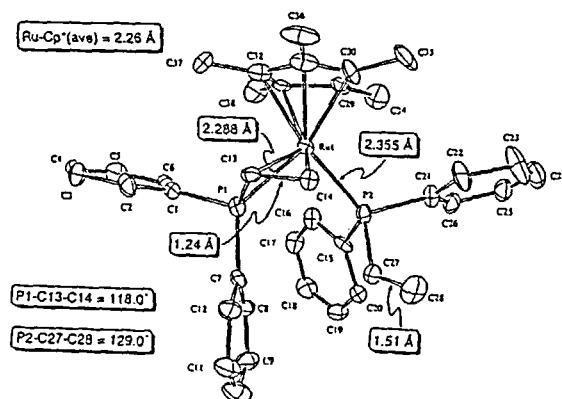
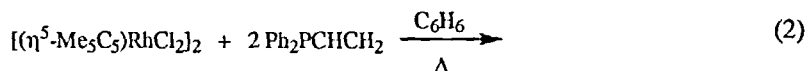
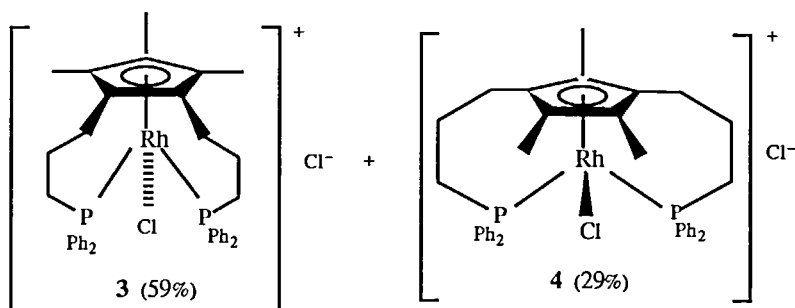


Figure 1. Structure of the cation of 1

The  $^{31}\text{P}\{^1\text{H}\}$  NMR spectrum exhibited three resonances:  $\delta$  - 145.0 (septet,  $^1\text{J}(\text{PF}) = 7$  12.7 Hz,  $\text{PF}_6^-$ ), 12.9 (d,  $^2\text{J}(\text{PP}) = 48.7$  Hz,  $\eta^3$  -  $\text{Ph}_2\text{PCHCH}_2$ ), 43.3. (d,  $^2\text{J}(\text{PP}) = 48.7$  Hz,  $\eta^1$  -  $\text{Ph}_2\text{PCHCH}_2$ ) rather than the two expected for the  $\text{CH}_3\text{CN}$  adduct. Thus, the  $\eta^1$ -to- $\eta^3$  conversion is spontaneous even in  $\text{CH}_3\text{CN}$  solution. X-ray crystallography and  $^1\text{H}$  NOE difference spectroscopy established that the phosphaallyl moiety adopts the *exo*- configuration in both the solid and solution states. Reaction 1 with  $\text{Ph}_2\text{PCH}_2\text{CHCH}_2$  in place of  $\text{Ph}_2\text{PCHCH}_2$  produced the  $\eta^3$ -homophosphaallyl analog  $[(\eta^5\text{-Me}_5\text{C}_5)\text{Ru}(\eta^1\text{-Ph}_2\text{PCH}_2\text{CHCH}_2)(\eta^3\text{-Ph}_2\text{PCH}_2\text{CHCH}_2)]^+ \text{PF}_6^-$  (2) in high yield. Compound (2) is dynamic in solution as shown by variable temperature  $^{31}\text{P}\{^1\text{H}\}$  NMR spectroscopy.

Analogous reactions of  $[(\eta^5\text{-Me}_5\text{C}_5)\text{RhCl}_2]_2$  with  $\text{Ph}_2\text{PCHCH}_2$  proceed quite unexpectedly<sup>2</sup> (reaction 2).





The structures of the two novel products of reaction 2 are shown in Figure 2.

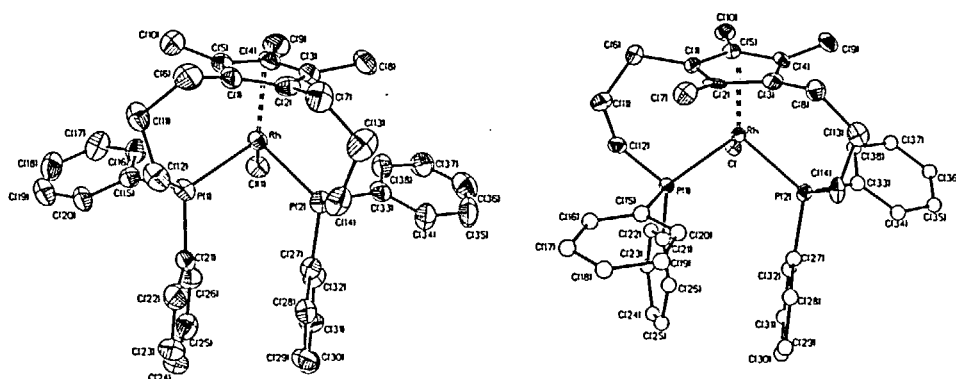


Figure 2. Structures of the cations of **3** (left) and **4** (right).

In contrast, reaction of  $[(\eta^5\text{-Me}_5\text{C}_5)\text{RhCl}_2]_2$  with  $\text{Ph}_2\text{PCH}_2\text{CHCH}_2$  and  $\text{NaPF}_6$  gave the ligand substitution product  $[(\eta^5\text{-Me}_5\text{C}_5)\text{Rh}(\text{Ph}_2\text{PCH}_2\text{CHCH}_2)_2\text{Cl}]^+ \text{PF}_6^-$  (**5**, Figure 3).

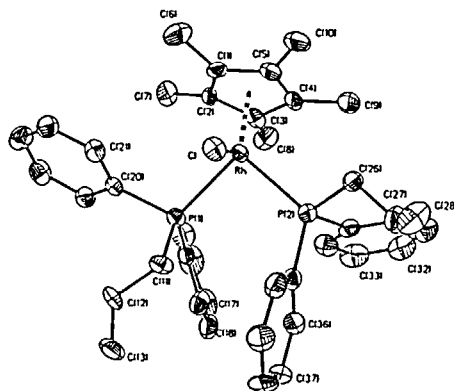


Figure 3. Structure of the cation of **5**.

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